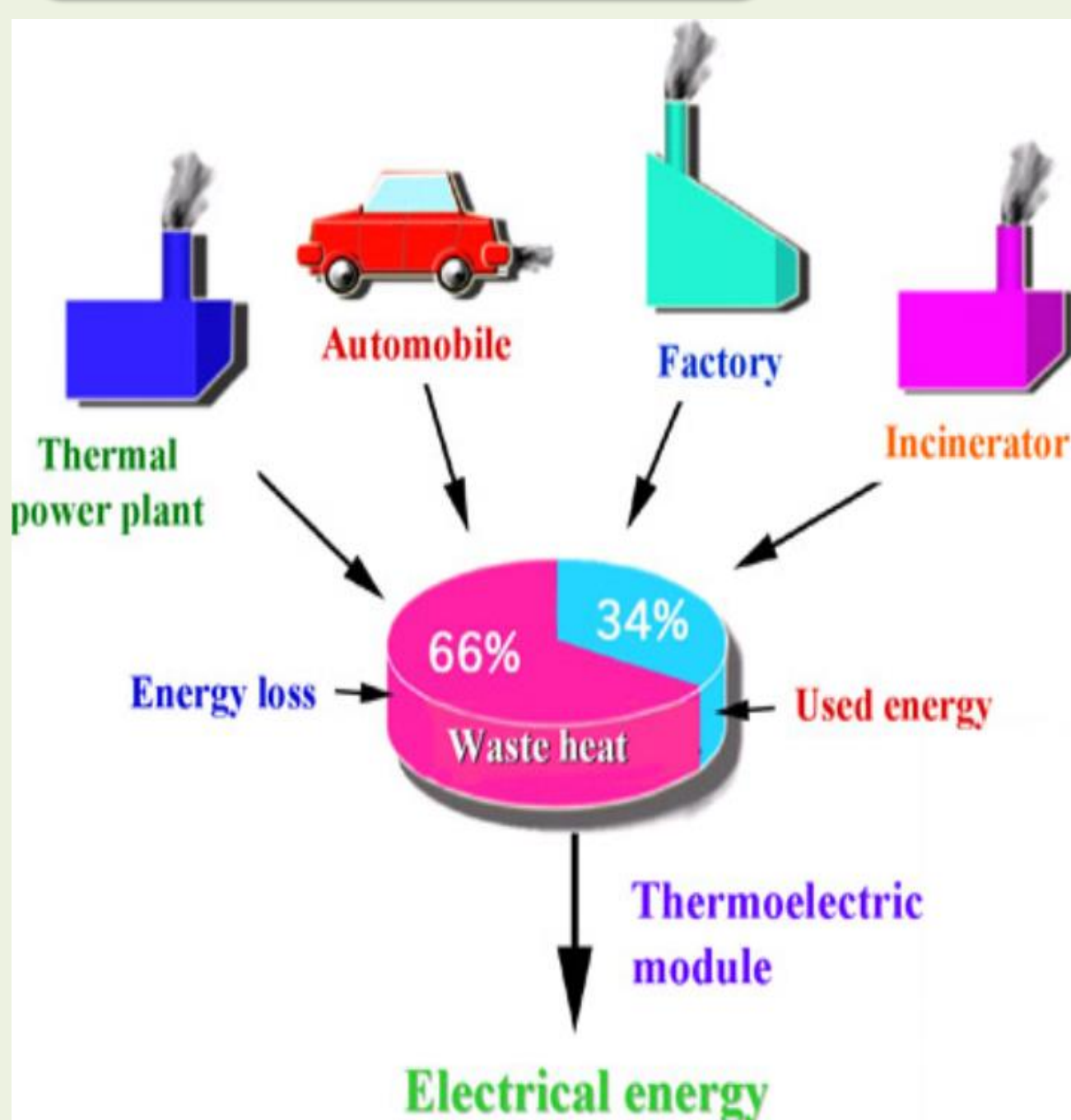


# Lanthanum cobaltite as a metallic oxide interconnect in thermoelectrics

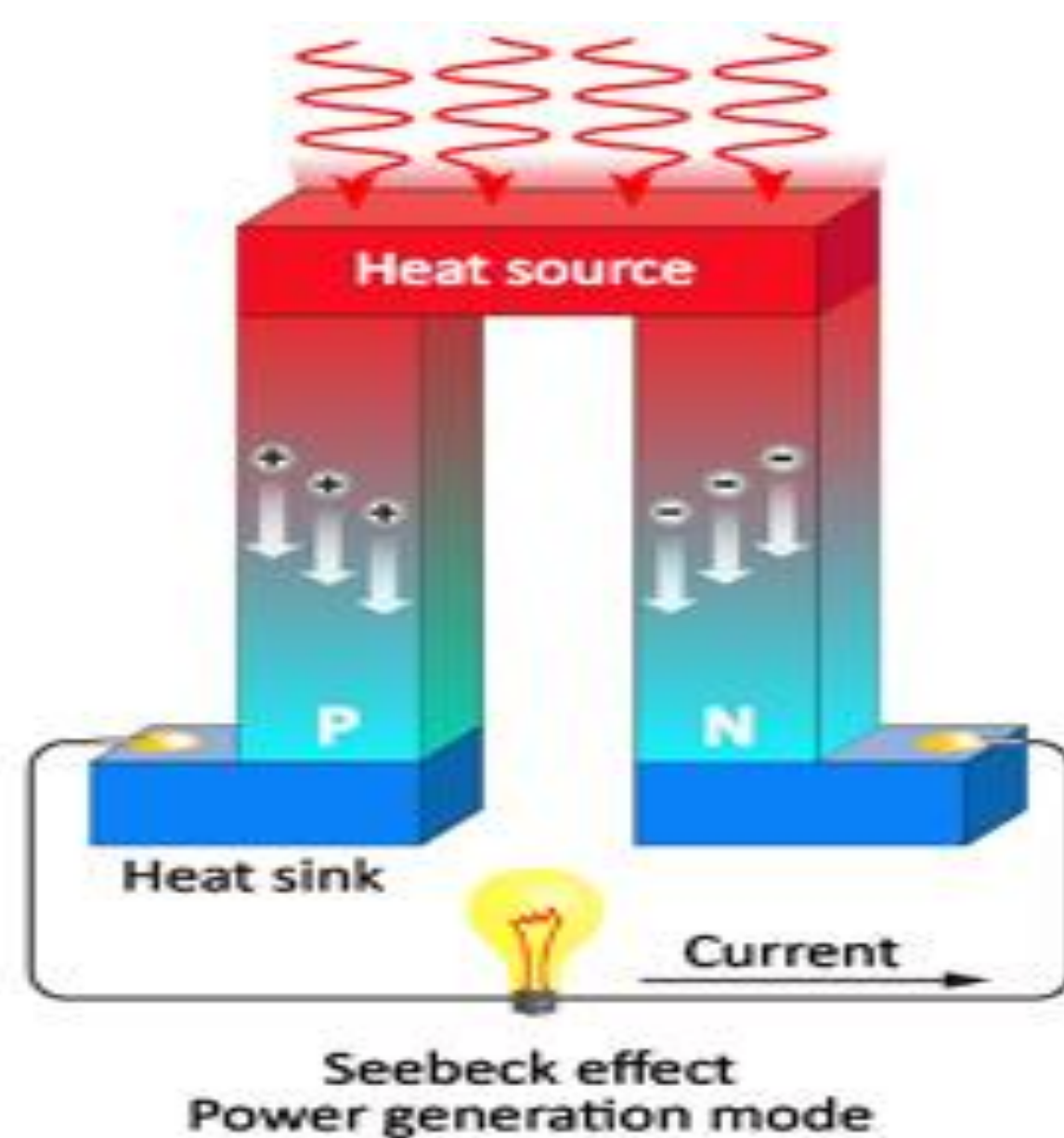
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## 1. Introduction



In 1821, Thomas Seebeck found that an electric current would flow continuously in a closed circuit made up of two dissimilar metals, if the junctions of the metals were maintained at two different temperatures.



### 1. Figure of Merit

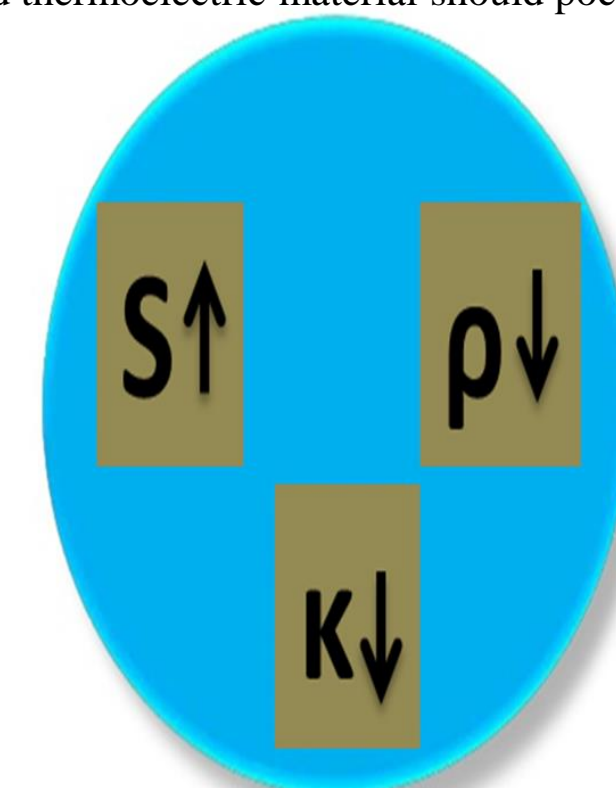
S, Seebeck coefficient,  
ρ, Electrical resistivity and  
κ, thermal conductivity

$$ZT = \frac{S^2 T}{\rho \kappa}$$

A good thermoelectric material should possess

### 2. Power factor

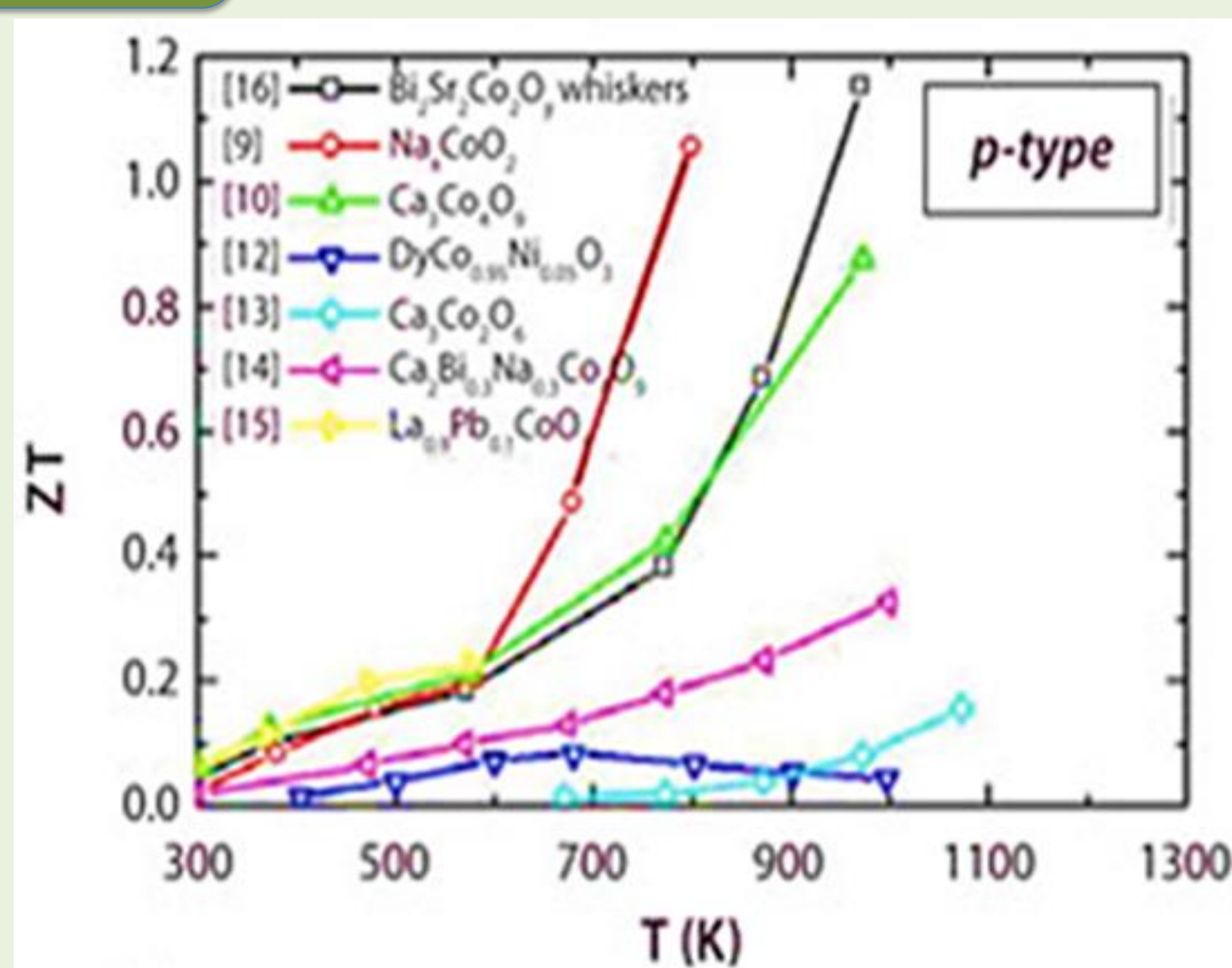
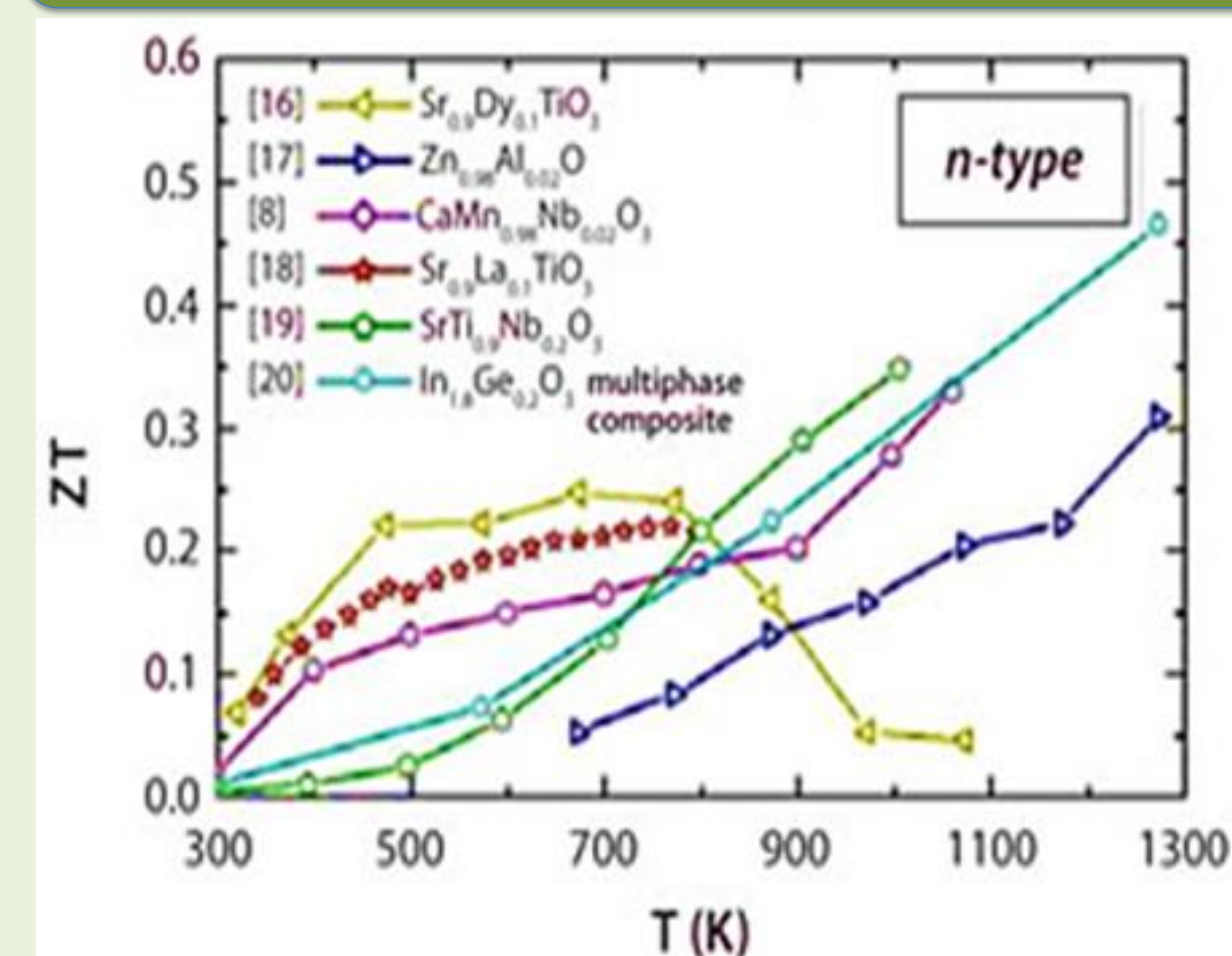
Power factor =  $\sigma S^2$   
σ, Electrical conductivity



### 3. Device efficiency

$\eta$  = Energy provided to the load / Heat energy absorbed at the hot junction

## 2. Oxide thermoelectric materials



- ✓ Low κ
- ✓ High S
- ✓ High thermal and chemical stability
- ✓ Environment friendly
- ✓ Abundant

**Main Challenge : Lower ZT range**

Matching the coefficient of thermal expansion

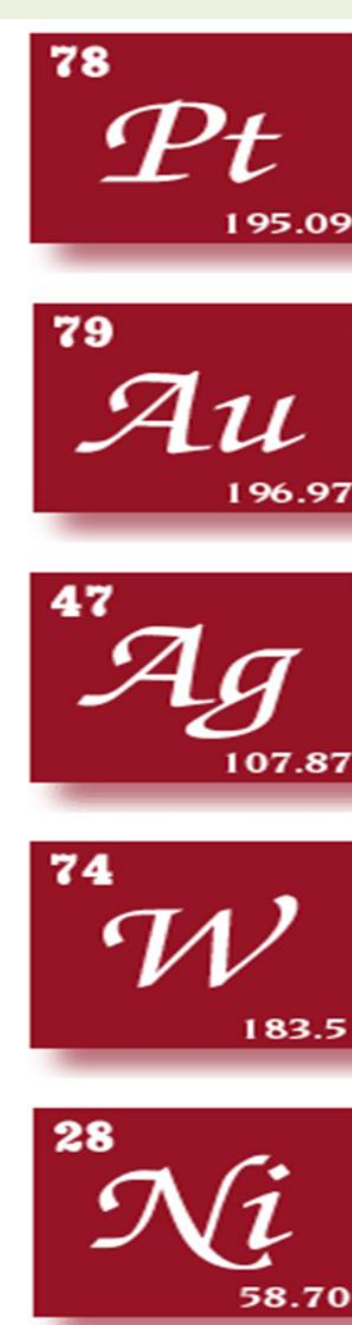
Bonding and interface reaction

Specific contact resistivity

- ❖ Metal diffuse at high temperature
- ❖ High contact resistance at the junction
- ❖ Cracking/ evaporation of metal contact
- ❖ High cost

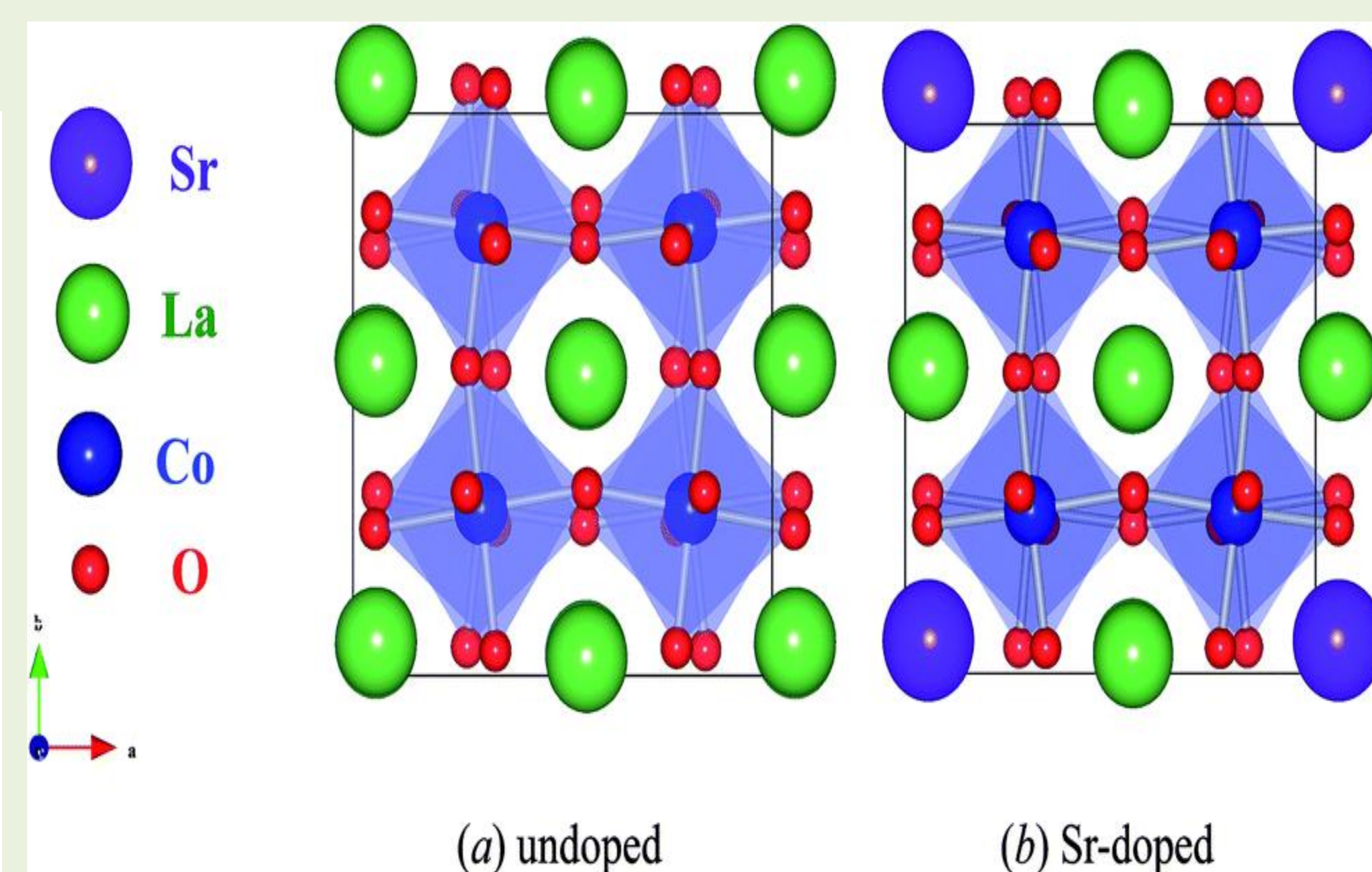
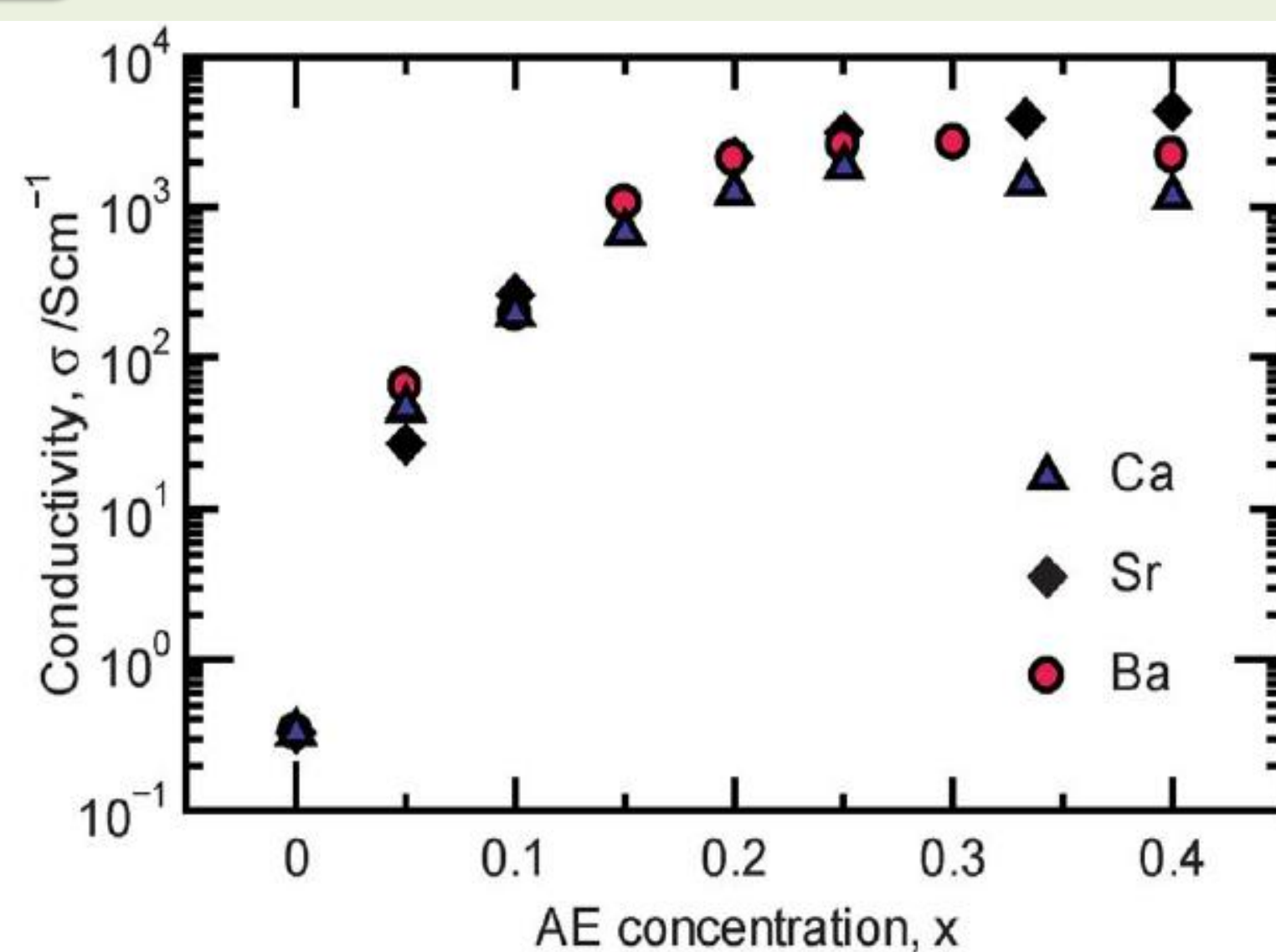
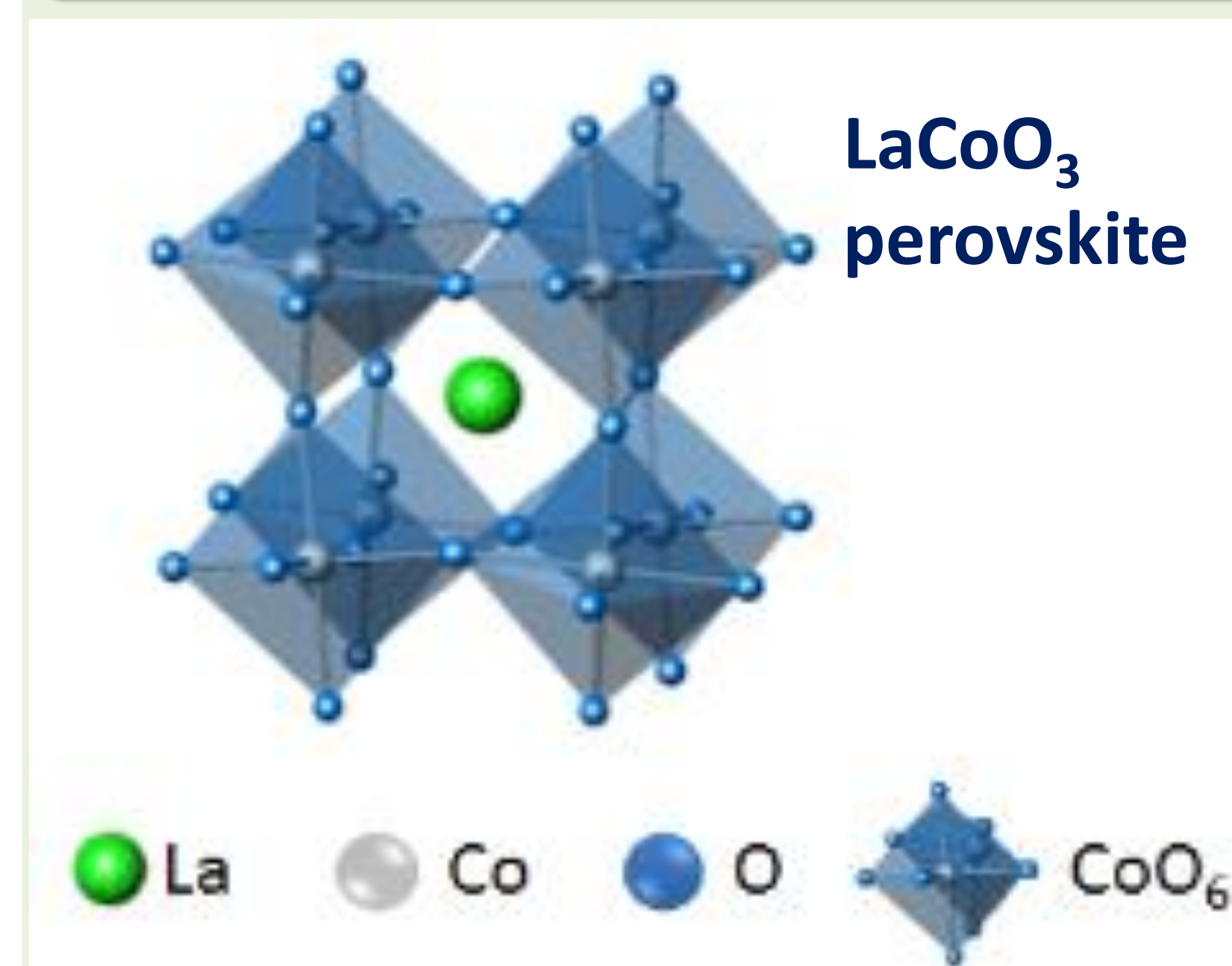
**Conducting oxides**

- High thermal stability
- High resistance to chemical corrosion
- Favorable integration with TE oxide



*p-n oxide thermoelectric junctions still need an ohmic contact: Noble metal? Or metallic oxide?  
In this project, we will investigate alkaline earth doped LaCoO<sub>3</sub> as metallic oxide interconnect.*

## 3. Alkaline earth doped LaCoO<sub>3</sub>



Electrical conductivity has a maximum,  $\sigma = 4.4 \times 10^3 \text{ S cm}^{-1}$  in La<sub>0.6</sub>Sr<sub>0.4</sub>CoO<sub>3</sub>

- High, metallic conductivity
- Low cost
- High thermal stability

## 4. References

1. Minnich, A.J., et al., *Energy & Environmental Science*, **2009**. 2(5) 466-479.
2. Fergus, J.W., *Journal of the European Ceramic Society*, **2012**. 32(3) 525-540.
3. Petrov, A.N., et al., *Solid State Ionics*, **1995**. 80(3) 189-199.